### FORMULARY REDUCER II FOR NEGATIVES

#### 1-liter kit

The action of Reducer II is super-proportional; it removes more silver metal from the highlights than it does from the shadow areas. Using a reducer correctly is an art and requires experience. We strongly urge you to practice with this reducer using scrap negatives before attempting reduction of a negative of value.

## CHEMICALS CONTAINED IN THIS KIT

Your kit contains the following chemicals.

<u>Chemical</u>	<u>Amount</u>
ammonium persulfate sulfuric acid, 50%	25 g 6 ml
sodium sulfite	50 g

### CHEMICAL SAFETY

All chemicals are dangerous and must be treated with respect. Please read the warning on each package. Two chemicals in this kit require special attention; sulfuric acid and ammonium persulfate.

<u>Sulfuric Acid</u>, <u>48%</u>. Sulfuric acid, even at 48% strength, is is a strong acid. If you should spill the acid, wash the area (or skin or clothing) immediately with copious amounts of cold water followed by soap and water. Neutralization of spilled acid with alkali is no longer a recommended clean-up procedure.

Ammonium Persulfate, an oxidizer, can supply oxygen to any combustible compound and, thus, is a potential fire hazard. Always clean up any spilled oxidizer with of water. Never dispose of excess solid oxidizer in a wastepaper basket -- rather, wash it down a sink drain with water.





# **Material Safety Data Sheet**

NFPA	HMIS		Personal Protective Equipment
(1)	Health Hazard	(3)	
1 0	Fire Hazard	(1)	
	Reactivity	0	See Section 15.

Common Name/	Ammonium persulfate	Code	A5220
Trade Name		CAS#	7727-54-0
Manufacturer	SPECTRUM CHEMICAL MFG. CORP.	RTECS	SE0350000
	14422 SOUTH SAN PEDRO STREET	TSCA	On the TSCA list
•	GARDENA, CALIFORNIA 90248-9985		
Commercial Name(s)	Not available	CI#	Not available.
Synonym	Ammonium peroxydisulfate	In case of emergency	
Chemical Name	Not available.	CHEMTREC (24hr) 800-424-9300	
Chemical Family	Not available.	-	
Chemical Formula	(NH4)2S2O8	Emergency p	hone: (310) 516-8000
Supplier	SPECTRUM QUALITY PRODUCTS 14422 S. SAN PEDRO STREET GARDENA, CA 90248-9985		

			Exposure Limits		i i
Name	CAS#	TWA (mg/m3)	STEL (mg/m3)	CEIL (mg/m3)	% by Weigh
Ammonium persulfate	7727-54-0				100

Toxicological Data on Ingredients

Ammonium persulfate:

ORAL (LD50):

Acute: 820 mg/kg (Rat).

## Section 3. Hazards Identification

Potential Acute Health Effects Extremely dangerous in case of eye contact (irritant). Very dangerous in case of skin contact (irritant), of ingestion, of inhalation. Slightly dangerous to dangerous in case of skin contact (corrosive). Very slightly to slightly dangerous in case of skin contact (permeator). Corrosive to eyes and skin. This product is a severe eye irritant. Prolonged exposure may result in skin burns and ulcerations. Over-exposure by inhalation may cause respiratory irritation. This product may irritate eyes and skin upon contact. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Potential Chronic CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. The substance is toxic to lungs, mucous membranes. Toxicity of the

Section 4. First Aid Measures

product to the reproductive system: Not available. Repeated or prolonged exposure to the substance can produce target organs damage.

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Check for and remove any contact lenses. IMMEDIATELY flush eyes with running water for at least 15 minutes, keeping eyelids open. COLD water may be used. DO NOT use an eye ointment. Seek medical attention.
If the chemical got onto the clothed portion of the body, remove the contaminated clothes as quickly as possible, protecting your own hands and body. Place the victim under a deluge shower. If the chemical touches the victim's exposed skin, such as the hands: Gently and thoroughly wash the contaminated skin with running water and non-abrasive soap. Be particularly careful to clean folds, crevices, creases and groin. COLD water may be used. Cover the irritated skin with an emollient. If irritation persists, seek medical attention. Wash contaminated clothing before reusing.
Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek medical attention.
Allow the victim to rest in a well ventilated area. Seek immediate medical attention.
Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. WARNING: It may be dangerous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled material is toxic, infectious or corrosive. Seek medical attention.
Remove dentures if any. Watch for an obstruction in the victim's mouth. Remove if possible what is causing the obstruction but do not force fingers or a hard object between the victim's teeth. Have conscious person drink several glasses of water or milk. INDUCE VOMITING by sticking finger in throat. Seek immediate medical attention.
Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive. Remove dentures if any. Watch for an obstruction in the victim's mouth. Remove if possible what is causing the obstruction but do not force fingers or a hard object between the victim's teeth. If a soft pad can be inserted between the victim's teeth, it will protect the tongue from being bitten. A badly bleeding tongue immensely complicates the patient's problems. Have conscious person drink several glasses of water or milk. INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. If convulsions occur, do not restrain the victim, but do remove objects with which he (she) might injure himself (herself) or orient the victim to prevent him (her) from striking fixed heavy objects. If the convulsions cease, turn the victim on the side or face down so that any fluid in the mouth will drain. Seek medical attention.

Section 5. Fire and Explosion Data	· ·			
Flammability of the Product	Combustible.			
Auto-Ignition Temperature	Not available.	 ,		
Flash Points	Not available.		· ·	
Flammable Limits	Not available.			

Products of Combustion	Not applicable.
Fire Hazards in Presence of Various Substances	Slightly flammable to flammable in presence of reducing materials.
Explosion in Presence of Various Substances	Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available. No specific information is available in our database regarding the product's risks of explosion in the presence of various materials.
Fire Fighting Media and Instructions	Oxidizing material. DO NOT use water jet. Cool containing vessels with water jet in order to prevent pressure build-up, autoignition or explosion. Use flooding quantities of water. Avoid contact with organic materials.
Special Remarks on Fire Hazards	No additional remark.
Special Remarks on Explosion Hazards	No additional remark.

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Small Spill	Use appropriate tools to put the spilled solid in a convenient waste disposal container.
	Oxidizing material. Stop leak if without risk. Avoid contact with a combustible material (wood,
Large Spill	paper, oil, clothing). Keep substance damp using water spray. DO NOT touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all sources of ignition. Call for assistance on disposal.

	containing material. DO NOT ingest. DO NOT breathe dust. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes.
Storage	Keep container dry. Keep in a cool place. Ground all equipment containing material. Oxidizing materials should be stored in a separate safety storage cabinet or room.

Engineering Controls	Use process enclosures, local exhaust ventilation, or other engeneering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.
Personal Protection	Splash goggles. Lab coat. Dust respirator. Be sure to use a MSHA/NIOSH approved respirator or equivalent. Gloves (impervious). Wear appropriate respirator when ventilation is inadequate.

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Protection in  Case of a sufficient; cons  Large Spill	sult a specialist BEFORE handling this product.	
Exposure Not available.		
Limits		
		1
Section 9. Physical and Chemica	I Properties	٦
Physical state	Solid. Odor Not available.	1
and appearance	With	
Molecular Weight	228.2	
pH (1% soln/water)	Not available.	
Boiling Point	Not available.	- -
Melting Point	Decomposes.	-
Critical Temperature	Not available.	-
		_
Specific Gravity	1.98 (Water = 1)	_
Vapor Pressure	Not available.	
Vapor Density	7.9 (Air = 1)	
Volatility	Not available.	
Odor Threshold	Not available.	
Evaporation rate	Not available.	_
Viscosity	Not available.	-
Water/Oil Dist. Coeff.	Not available.	1
Ionicity (in Water)	Not available.	1
Dispersion Properties	See solubility in water.	-
Solubility	Easily soluble in cold water. Soluble in hot water.	-
Section 10. Stability and Reactive	ity Data	Ī
Stability	The product is stable.	7
Instability Temperature	Not available.	-
Conditions of Instability	No additional remark.	-
Incompatibility with	No specific information is available in our database regarding the reactivity	-
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Section 10. Stability and Reactivi	ty Data
Stability	The product is stable.
Instability Temperature	Not available.
Conditions of Instability	No additional remark.
Incompatibility with Various Substances	No specific information is available in our database regarding the reactivity of this material in presence of various other materials.
Corrosivity	Non-corrosive in presence of glass.
Cuttosivity	redirection of presence of glass.
Special Remarks on Reactivity	No additional remark.
Special Remarks on Corrosivity	No additional remark.

Ī	Section 11. Toxicological Information
Ī	Routes of Entry Ingestion. Inhalation.

Toxicity to Animals	Acute oral toxicity (LD50): 820 mg/kg (Rat).
e .	
Chronic Effects on Humans	The substance is toxic to lungs, mucous membranes. Toxicity of the product to the reproductive system: Not available.
Other Toxic Effects on Humans	Extremely dangerous in case of eye contact (irritant). Very dangerous in case of skin contact (irritant), of ingestion, of inhalation. Slightly dangerous to dangerous in case of skin contact (corrosive). Very slightly to slightly dangerous in case of skin contact (permeator).
Special Remarks on Toxicity to Animals	No additional remark.
Special Remarks on Chronic Effects on Humans	No additional remark.
Special Remarks on other Toxic Effects on Humans	No additional remark.

Ecotoxicity	Not available.	
BOD5 and COD	Not available.	
Products of	Not applicable.	
Biodegradation		
Toxicity of the	The products of degradation are more toxic.	
Products of	•	
Biodegradation		
Special Remarks on	No additional remark.	
the Products of		
Biodegradation		

Waste Disposal	Recycle to process,	if possible. Consult	your local or region	onal authorities.	
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	en en en				
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DOT Classification	DOT CLASS 5.1: Oxidizer.		
Identification	Ammonium Persulfate UN1444 III	-	
Special Provisions	No additional remark.		
for Transport	elisa di Salaharan		
DOT (Pictograms)			-
			<u> </u>
	OXIDIZER		

## Federal and State Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA) Title III requires emergency planning based on Threshold Planning Quantities (TPQs) and release reporting based on Reportable Quantities (RQs) in 40 CFR 355 (used for SARA 302, 304, 311 and 312). Components present in this product at a level which could require reporting under the statute are:

#### NONE

Superfund Amendments and Reauthorization Act of 1986 (SARA) Title III requires submission of annual report release of toxic chemicals that appear in 40 CFR 372 (used for SARA 313). This information must be included in all MSDSs that are copied and distributed for this material. Components present in this product at a level which could require reporting under the statute are:

#### NONE

Pensylvania Right-To-Know, Hazardous substance List, Hazardous Substances and Special hazardous Substances on the list must be identified when present in products. Components present in this product at a level which could require reporting under the statute are:

#### NONE

Massachusetts Right-To-Know, Substance List (MSL) Hazardous Substances and Extraordinarily Hazardous Substances on the MSL must be identified when present in products. Components present in this product at a level which could require reporting under the statute are:

### NONE

Comprehensive Environmental Response, Compensation, and Liability Act of 1980 (CERCLA) requires notification of the National Response Center of release of quantities of Hazardous Substances equal or greater than the reportable quantities (RQs) in 40 CFR 302.4. Components present in this product at a level which could require reporting under the statute are:

### NONE

WARNING: This product contains a chemical known to the State of California to cause cancer. Chemical ingredient(s) requiring this warning:

## NONE

WARNING: This product contains a chemical known to the State of California to cause birth defects or other reproductive harm. Chemical ingredient(s) requiring this warning: NONE

Other Classifications

WHMIS (Canada)

WHMIS CLASS C: Oxidizing material. WHMIS CLASS D-2A: Material causing other toxic effects (VERY TOXIC).

DSCL (EEC)

R8- Contact with combustible material may cause fire.

R9- Explosive when mixed with combustible material.

Other Classifications WHMIS (Canada) WHMIS CLASS C: Oxidizing material. WHMIS CLASS D-2A. Material causing other toxic effects (VERY TOXIC). DSCL (EEC) R8- Contact with combustible material may cause fire. R9- Explosive when mixed with combustible material. R22- Harmful if ingested. R41- Risk of serious damage to eyes. R42- May cause sensitization by inhalation. Health Hazard HMIS (U.S.A.) National Fire Protection Association (U.S.A.) Fire Hazard Flammability Reactivity Health Reactivity Personal Protection Specific hazard

## Personal Protective Equipment



Protective Gloves (impervious).



Lab coat.



Dust respirator. Be sure to use a MSHA/NIOSH approved respirator or equivalent. Wear appropriate respirator when ventilation is inadequate.



Splash goggles.

References	Not available.	
		•
	·	
Catalog Number(s)	A1225, A1227, A1230	
Other Special	No additional remark.	
Considerations		
Validated by	E. Bruil on 12/17/96.	Verified by E. Brull.
		Name
Emergency Phone:	(310)516-8000	
Notice to Reader	All chamicals may page unknow	vn hazards and should be used with caution. This Material
		with nazards and should be used with caution. This Material biles only to the material as packaged. If this product
	is combined with other material	s, deteriorates, or becomes contaminated, it may pose
	hazards not mentioned in this N	ISDS. It shall be the user's responsibility to develop proper
	methods of handling and persor	nal protection based on the actual conditions of use. While

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PHCTOGRAPHERS FORMULARY INC PO BOX 5165 MISSCULA, MT 59806

ORDER NO: PROD NO: 04722007

VAN WATERS & ROGERS INC. 1600 NORTON BLDG. SEATT	LE, WA 98104-1564				
EMERGENCY ASSISTANCE	e para de la composição d				
FOR EMERGENCY ASSISTANCE INVOLVING CHEMICALS CALI (800) 424-9300.	CHEMTREC				
FOR PRODUCT AND SALES INFORMATION	the flesh fit is first from the first from their deem days flesh flesh flesh grain grain same				
CONTACT YOUR LOCAL VAN WATERS & ROGERS BRANCH					
PRODUCT IDENTIFICATION					
PRODUCT NAME: SULFURIC ACID  COMMON NAMES/SYNONYMS: SULFURIC ACID;  COF VITRIOL; SULFURIC ACID > 93%; SULFURIC ACID > 66  FURIC ACID > 1.83 SPECIFIC GRAVITY	: 7664-93-9 E: T1361 DEG BE;				
PRODUCT NAME: SULFURIC ACID  COMMON NAMES/SYNONYMS: SULFURIC ACID;  OF VITRIOL; SULFURIC ACID > 93%; SULFURIC ACID > 66  FURIC ACID > 1.83 SPECIFIC GRAVITY  FORMULA: H2 S O4  HA ZARD RATING (NFPA 49)  HEALTH: 3  FIRE: 0  REACTIVITY: 2  SPECIAL: NO WATER  CAS NO  DATE ISSUE  CAS NO  VW&R CODI  VW&R CODI  VW&R CODI  VW&R CODI  VW&R CODI  VWATER DATE  OATE ISSUE  SUPERCEDES  0=MINIMAL  1=SLIGHT  2=MODERATE	D: 07/87 S: 01/87 ING SCALE: 3=SERIOUS 4=SEVERE				
HAZARDOUS INGREDIENTS	nn dan dan dan dan dan dan dan dan dan d				
EXPOSURE LIMITS, MG/M3 OSHA ACGIH OTHER COMPONENT CAS NO. % PEL TLV LIMIT SULFURIC ACID 7664-93-9 >93 1 1 1 (DUPONT) WATER 7732-18-5 BALANCE NONE NONE NONE	HAZARD CORROSIVE NONE				
i .					
BOILING POINT, DEG F: A = 529; VAPOR PRESSURE, MM HG, $B = 590$ FREEZING POINT, DEG F: $A = -20$ ; VAPOR DENSITY	720 DEG C: A,B = NIL NIL (AIR=1): N/A				
B = 30 SPECIFIC GRAVITY (WATER=1): A = 1.835; WATER SOLUBILITY, %:					
SPECIFIC GRAVITY (WATER=1): A = 1.835; WATER SOLUTE B = 1.84  APPEARANCE AND ODOR: COLOR- EVAPORATION RATE (BUTYL ACLESS TO PALE YELLOW, OILY LIQUID. ODORLESS.	COMPLETE CETATE=1): <1				
A = 93% OR 66 DEG BE SULFURIC ACID; B = 99% SULFURIC	ACID				
FIRST AID MEASURES					
INHALED: REMOVE TO FRESH AIR. GIVE ARTIFICIAL RESPI	RATION IF NOT				
IN CASE OF EYE CONTACT: IMMEDIATELY FLUSH EYES WITH LOT	S OF RUNNING				
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CULFURIC ACID WATER WHITE

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WATER FOR 30 MINUTES, LIFTING THE UPPER AND LOWER EYELIDS OCCASIONALLY. GET IMMEDIATE MEDICAL ATTENTION.

IN CASE OF SKIN CONTACT: IMMEDIATELY FLUSH SKIN WITH LOTS OF RUNNING WATER FOR 30 MINUTES. REMOVE CONTAMINATED CLOTHING AND SHOES; WASH BEFORE REUSE. GET IMMEDIATE MEDICAL ATTENTION.

IF SWALLOWED: DO NOT INDUCE VOMITING. IF CONSCIOUS, GIVE LOTS OF WATER OR MILK. GET IMMEDIATE MEDICAL ATTENTION. DO NOT GIVE ANYTHING BY MOUTH TO AN UNCONSCIOUS OR CONVULSING PERSON.

-----HEALTH HAZARD INFORMATION-----

PRIMARY ROUTES OF EXPOSURE: SKIN OR EYE CONTACT

SIGNS AND SYMPTOMS OF EXPOSURE
INHALATION: VAPORS AND MISTS ARE EXTREMELY CORROSIVE TO THE NOSE,
THROAT, AND MUCOUS MEMBRANES. BRONCHITIS, PULMONARY EDEMA, AND
CHEMICAL PNEUMONITIS MAY OCCUR. IRRITATION, COUGHING, CHEST PAIN, AND
DIFFICULTY IN BREATHING MAY OCCUR WITH BRIEF EXPOSURE WHILE PROLONGED
EXPOSURE MAY RESULT IN MORE SEVERE IRRITATION AND TISSUE DAMAGE.
BREATHING HIGH CONCENTRATIONS MAY RESULT IN DEATH.

EYE CONTACT: VAPORS, LIQUID, AND MISTS ARE EXTREMELY CORROSIVE TO THE EYES. BRIEF CONTACT OF THE VAPORS WILL BE SEVERELY IRRITATING. BRIEF CONTACT OF THE LIQUID OR MISTS WILL SEVERELY DAMAGE THE EYES AND PROLONGED CONTACT MAY CAUSE PERMANENT EYE INJURY WHICH MAY BE FOLLOWED BY BLINDNESS.

SKIN CONTACT: VAPORS, MISTS, AND LIQUID ARE EXTREMELY CORROSIVE TO THE SKIN. VAPORS WILL SEVERELY IRRITATE THE SKIN AND LIQUID AND MISTS WILL SEVERELY BURN THE SKIN. PROLONGED LIQUID CONTACT WILL BURN DESTROY SURROUNDING TISSUE AND DEATH MAY ACCOMPANY BURNS WHICH EXEND OVER LARGE PORTIONS OF THE BODY.

SWALLOWED: VAPORS, MISTS, AND LIQUID ARE EXTREMELY CORROSIVE TO THE MOUTH AND THROAT. SWALLOWING THE LIQUID BURNS THE TISSUES, CAUSES SEVERE ABDOMINAL PAIN, NAUSEA, VOMITING, AND COLLAPSE. SWALLOWING LARGE QUANTITIES CAN CAUSE DEATH.

CHRONIC EFFECTS OF EXPOSURE: MAY CAUSE EROSION OF THE TEETH, LESIONS ON THE SKIN, BRONCHIAL IRRITATION, COUGHING, AND PNEUMONIA.

MEDICAL CONDITIONS GENERALLY AGGRAVATED BY EXPOSURE: ACUTE AND CHRONIC RESPIRATORY DISEASES.

TOXICITY DATA----

ORAL: RAT LD50 = 2.140 MG/KG

DERMAL: NO DATA FOUND

INHALATION: GUINEA PIG LC50 = 18 MG/M3

CARCINOGENICITY: THIS MATERIAL IS NOT CONSIDERED TO BE A CARCINOGEN BY THE NATIONAL TOXICOLOGY PROGRAM, THE INTERNATIONAL AGENCY FOR RESEARCH ON CANCER, OR THE OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION

OTHER DATA: ALTHOUGH ONE LIMITED STUDY OF REFINERY WORKERS DID SUGGEST A POSSIBLE LINK BETWEEN SULFURIC ACID EXPOSURE AND LARYNGEAL CANCER, THE STUDY WAS LIMITED BECAUSE OF THE SMALL NUMBER OF WORKERS AND THE MIXED EXPOSURES TO SEVERAL OTHER MATERIALS INCLUDING DIETHYL SULFATE, AN IARC AND NTP CARCINOGEN. BASED ON THE OVERALL WEIGHT OF EVIDENCE FROM ALL ANIMAL TOXICITY AND HUMAN EPIDEMIOLOGICAL STUDIES, NO CAUSE-AND-EFFECT RELATIONSHIP BETWEEN CANCER AND SULFURIC ACID EXPOSURE HAS BEEN SHOWN. INDIVIDUALS WITH PREEXISTING DISEASE OF THE LUNGS MAY HAVE INCREASED USCEPTIBILITY TO THE TOXICITY OF EXCESSIVE EXPOSURES.

-- PERSONAL PROTECTION-----

VENTILATION: LOCAL MECHANICAL EXHAUST VENTILATION CAPABLE OF MAIN-

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INVOICE:

TILFURIC ACID WATER WHITE

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TAINING EMISSIONS AT THE POINT OF USE BELOW THE PEL.

RESPIRATORY PROTECTION: WEAR A NIOSH-APPROVED RESPIRATOR APPROPRIATE FOR THE VAPOR OR MIST CONCENTRATION AT THE POINT OF USE. APPROPRIATE RESPIRATORS MAY BE A FULL FACEPIECE AIR-PURIFYING CARTRIDGE RESPIRATOR EQUIPPED FOR ACID GASES/MISTS, A SELF-CONTAINED BREATHING APPARATUS IN THE PRESSURE DEMAND MODE, OR A SUPPLIED-AIR RESPIRATOR.

EYE PROTECTION: CHEMICAL GOGGLES AND FULL FACESHIELD UNLESS A FULL FACEPIECE RESPIRATOR IS ALSO WORN. IT IS GENERALLY RECOGNIZED THAT CONTACT LENSES SHOULD NOT BE WORN WHEN WORKING WITH CHEMICALS BECAUSE CONTACT LENSES MAY CONTRIBUTE TO THE SEVERITY OF AN EYE INJURY.

PROTECTIVE CLOTHING: ACID-RESISTANT SLICKER SUIT WITH RUBBER APRON, RUBBER BOOTS WITH PANTS OUTSIDE, AND RUBBER GLOVES WITH GAUNTLETS.

AN EYEWASH AND SAFETY SHOWER SHOULD BE OTHER PROTECTIVE MEASURES: NEARBY AND READY FOR USE.

FIRE AND EXPLOSION INFORMATION----

METHOD USED: N/A

EXTINGUISHING MEDIA: THIS MATERIAL IS NOT COMBUSTIBLE. USE EXTINGUISHING MEDIA APPROPRIATE FOR SURROUNDING FIRE.

SPECIAL FIRE BIOUZZIA ---FLASH POINT, DEG F: NONE
METHOD USED: N/A
EXTINGUISHING MEDIA: TH

SPECIAL FIRE FIGHTING PROCEDURES: FIRE FIGHTERS SHOULD WEAR SELF-CONTAINED BREATHING APPARATUS AND FULL PROTECTIVE CLOTHING. USE WESPRAY TO COOL NEARBY CONTAINERS AND STRUCTURES EXPOSED TO FIRE. USE WATER

NUSUAL FIRE AND EXPLOSION HAZARDS: EXTINGUISH ALL NEARBY SOURCES OF INITION SINCE FLAMMABLE HYDROGEN GAS WILL BE LIBERATED FROM CONTACT WITH SOME METALS. KEEP WATER OUT OF CONTAINERS.

\_\_\_\_HAZARDOUS REACTIVITY

STABLE CONDITIONS TO AVOID: NONE

POLYMERIZATION: WILL NOT OCCUR

MATERIALS TO AVOID: ALKALIS, OXIDIZING OR REDUCING MATERIALS, CYANIDES, SULFIDES, OR COMBUSTIBLE MATERIALS. REACTS WITH MANY METALS. CON-CENTRATED ACID REACTS VIOLENTLY WITH WATER.

HAZARDOUS DECOMPOSITION PRODUCTS: MAY LIBERATE CARBON MONOXIDE, CARBON DIOXIDE, AND OXIDES OF SULFUR.

-----SPILL, LEAK, AND DISPOSAL PROCEDURES

ACTION TO TAKE FOR SPILLS OR LEAKS: WEAR ACID-RESISTANT SLICKER SUIT AND COMPLETE PROTECTIVE EQUIPMENT INCLUDING RUBBER GLOVES, RUBBER BOOTS, AND A SELF-CONTAINED BREATHING APPARATUS IN THE PRESSURE DEMAND MODE OR A SUPPLIED-AIR RESPIRATOR. IF THE SPILL OR LEAK IS SMALL, A FULL FACE-PIECE AIR-PURIFYING CARTRIDGE RESPIRATOR EQUIPPED FOR ACID GASES MAY BE SATISFACTORY. IN ANY EVENT, ALWAYS WEAR EYE PROTECTION. REMOVE ALL SOURCES OF IGNITION. FOR SMALL SPILLS OR DRIPS, MOP OR WIPE UP AND DISPOSE OF IN DOT-APPROVED WASTE CONTAINERS. FOR LARGE SPILLS, CONTAIN BY DIKING WITH SOIL OR OTHER NON-COMBUSTIBLE ABSORBENT MATERIAL AND CARE-FULLY NEUTRALIZE WITH SODA ASH OR LIME. IF SODA ASH IS USED, PROVIDE ADEQUATE VENTILATION TO DISSIPATE THE CARBON DIOXIDE GAS. KEEP NON-NEUTRALIZED MATERIAL OUT OF SEWERS, STORM DRAINS, SURFACE WATERS, AND SOIL.

SOIL.
COMPLY WITH ALL APPLICABLE GOVERNMENTAL REGULATIONS ON SPILL REPORTING,
AND HANDLING AND DISPOSAL OF WASTE.

PISPOSAL METHODS: DISPOSE OF CONTAMINATED PRODUCT AND MATERIALS USED N CLEANING UP SPILLS OR LEAKS IN A MANNER APPROVED FOR THIS MATERIAL. CONSULT APPROPRIATE FEDERAL, STATE AND LOCAL REGULATORY AGENCIES TO ASCERTAIN PROPER DISPOSAL PROCEDURES.
NOTE: EMPTY CONTAINERS CAN HAVE RESIDUES, GASES AND MISTS AND ARE SUBJECT TO PROPER WASTE DISPOSAL, AS ABOVE.

INVOICE:

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# SPECIAL PRECAUTIONS----

STORAGE AND HANDLING PRECAUTIONS: STORE IN A COOL, DRY, WELL-VENTILATED PLACE AWAY FROM INCOMPATIBLE MATERIALS. VENT CONTAINER CAREFULLY, AS NEEDED, TO RELIEVE PRESSURE. KEEP CONTAINER TIGHTLY CLOSED WHEN NOT IN USE. DO NOT USE PRESSURE TO EMPTY CONTAINER. WASH THOROUGHLY AFTER HANDLING. DO NOT GET IN EYES, ON SKIN, OR ON CLOTHING.

REPAIR AND MAINTENANCE PRECAUTIONS: DO NOT CUT, GRIND, WELD, OR DRILL ON OR NEAR THIS CONTAINER.

OTHER PRECAUTIONS: CONTAINERS, EVEN THOSE THAT HAVE BEEN EMPTIED, WILL RETAIN PRODUCT RESIDUE AND VAPORS. ALWAYS OBEY HAZARD WARNINGS AND HANDLE EMPTY CONTAINERS AS IF THEY WERE FULL.

.....FOR ADDITIONAL INFORMATION----

CONTACT DOUGLAS EISNER, TECHNICAL DIRECTOR, VAN WATERS & ROGERS INC. DURING BUSINESS HOURS, PACIFIC TIME (206)447-5911

---- NOTICE----

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07/87: REPORTED ADDITIONAL EXPOSURE LIMIT AND OTHER TOXICITY DATA.

PROD: 04722007 09:23:42 14 OCT 1988 CUST:

INVOICE:

SPORT NUMBER: 703

VAN WATERS & ROGERS INC.

MATERIAL SAFETY DATA SHEET

308 NO: 80003048 FFECTIVE DATE: 11/23/92

-.VERSION: 002

PAGE: 001

CODUCT: SODIUM SULFITE

ORDER NO: 141120 -PROD NO : 503217

PHOTOGRAPHERS FORMULARY C/O UNITED FRT, TERMINAL

KALISPELL ,MT 59806

AN WATERS & ROGERS INC. , SUBSIDIARY OF UNIVAR (206)889-3400 100 CARILLON POINT , KIRKLAND

-----EMERGENCY ASSISTANCE -------

FOR EMERGENCY ASSISTANCE INVOLVING CHEMICALS CALL - CHEMIREC (800)424-9300

CONTACT YOUR LOCAL VAN WATERS & ROGERS BRANCH OFFICE AT VW&R SPOKANE 509-534-0405 SPOKANE , WA

RODUCT NAME: DDIUM SULFITE

3DS #: GC003048

. GENERAL INFORMATION

RADE NAME (COMMON NAME): SODIUM SULFITE, SULFTECH(TM) grade Sodium Sulfite .A.S. No. 7757-83-7

TEMICAL NAME AND/OR SYNONYM: Sodium Sulfite

TRMULA: Na2SO3

DLECULAR WEIGHT: 126.04

DORESS: GENERAL CHEMICAL CORPORATION

90 East Halsey Road

Parsippany, NJ 07054-0389

CC CONTACT: Manager of Product Safety:

CC PHONE NUMBER: (201) 515-1840

JRRENT ISSUE DATE: July, 1990

. FIRST AID MEASURES

10 EMERGENCY PHONE NUMBER: (800) 431-8050

(ES: Immediately flush with plenty of water, for at least 15 minutes. Get

medical attention.

(IN: Promptly wash with plenty of soap and water.

MALATION: Remove to fresh air. If symptoms persist, get medical attention.

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INGESTION: If conscious, immediately give 2 to 4 glasses of water or milk and induce vomiting by touching finger to back of throat. Get immediate medical attention.

## C. HAZARDS INFORMATION

BHEALTH -

INHALATION: Inhalation of product dust or mist may irritate respiratory tract. Contact with acids liberates irritating and potentially fatal sulfur dioxide gas.

INGESTION: Ingestion may irritate gastrointestinal tract. Estimated to be moderately toxic. May cause severe allergic reaction in some asthmatics and sulfite sensitive individuals. Large doses may cause violent colic and diarrhea, circulatory disturbances, central nervous system depression and even death.—Reference (a).

SKIN: Dust or mist may cause skin irritation from prolonged contact.
Solutions will irritate. See pH, Section F. Rabbit and guinea pig data available, Reference (b).

EYES: Dust or mist may irritate or burn eyes. Solutions will irritate or burn. See  $pH_{\star}$  Section F.

PERMISSIBLE CONCENTRATION: AIR: (SEE SECTION J)

None established for sodium sulfite. OSHA/TWA for SO2 = 2 ppm OSHA/STEL for SO2 = 5 ppm BIOLOGICAL: None established.

UNUSUAL CHRONIC TOXICITY: See Section K.

FIRE AND EXPLOSION -

FLASH POINT: Not flammable. AUTO IGNITION TEMPERATURE: NA

FLAMMABLE LIMITS IN AIR (% BY VOL.): LOWER - NA UPPER - NA UNUSUAL FIRE AND EXPLOSION HAZARDS: See Hazardous Decomposition Pro

UNUSUAL FIRE AND EXPLOSION HAZARDS: See Mazardous Decomposition Products, Section G.

## D. PRECAUTIONS/PROCEDURES

FIRE EXTINGUISHING AGENTS RECOMMENDED: NA FIRE EXTINGUISHING AGENTS TO AVOID: NA

SPECIAL FIRE FIGHTING PRECAUTIONS: Wear NIOSH-approved self-contained breathing apparatus. Use water-spray to keep containers cool, and to knock down fumes.

VENTILATION: LOCAL EXHAUST if dusty or misty condition prevails.

LOCAL EXHAUST if there is release of sulfur dioxide gas, see Section G. Keep incompatible materials out of hoods, ducts, etc.

NORMAL HANDLING: Avoid contact with eyes, skin, clothing. Avoid breathing dust or mist. Use with adequate ventilation.

STORAGE: Store in a cool, dry area, away from acids or oxidizers.

Keep container closed. Protect from physical damage.

SPILL OR LEAK (ALWAYS WEAR PERSONAL PROTECTIVE EQUIPMENT - SECTION E):

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Promptly shovel up dry chemical into an empty container, and cover. Store as above. Cautiously spray residue with plenty of water. (See Section I for disposal methods.)

PECIAL: PRECAUTIONS/PROCEDURES/LABEL INSTRUCTIONS: SIGNAL WORD - WARNING: Contact with acids releases irritating and potentially fatal sulfur dioxide gas. See drum-handling instructions on label. When dissolving, add water cautiously and with stirring.

## . PERSONAL PROTECTIVE EQUIPMENT

ESPIRATORY PROTECTION: If dusty of misty conditions prevail, use dust or  $\cdot$ mist respirator approved by NIOSH. If sulfur dioxide should be released (see Section G), use respiratory protection approved by NIOSH for this gas. -- Reference (c).

(ES AND FACE: If exposed to dust or mist or solution, wear hard hat (or other head covering) and chemical safety goggles. Do not wear contact lenses.

ANDS, ARMS, AND BODY: Wear full work-clothing, including long-sleeved shirt and trousers for routine product-handling. Cotton gloves are usually adequate for dry product. For solutions, wear impervious gloves and apron. If contact is repeated or prolonged, wear full impervious clothing.

THER CLOTHING AND EQUIPMENT: Eyewash facility,

#### PHYSICAL DATA

TERIAL IS (AT NORMAL CONDITIONS): SOLID PEARANCE AND ODOR: White granular crystals or powder. Odorless. HILING POINT: Decomposes 900 Degrees C ILTING POINT: NO 'ECIFIC GRAVITY (H20=1): 2.63 POR DENSITY (AIR=1): NA MLUBILITY IN WATER (% by Weight): (Calculated as the anhydrous salt) 17% solution at 10 Degrees C

1: 1% solution; pH=9.8 (approx.) POR PRESSURE (mm Hg at 20 Degrees C): NA 'APORATION RATE (Ether=1): NA (Butyl Acetate = 1): NA VOLATILES BY VOLUME (At 20 Degrees C): NA

## REACTIVITY DATA

ABILITY: STABLE

INDITIONS TO AVOID: High temperature (before melting); yield sulfur dioxide gas and hazardous residue (details below).

28% solution at 33.4 Degrees C

COMPATIBILITY (MATERIALS TO AVOID): STRONG OXIDIZERS cause vigorous exothermic reactions.

ACIDS release sulfur dioxide gas (details below).

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HAZARDOUS DECOMPOSITION PRODUCTS: SULFUR DIOXIDE GAS: toxic and corrosive.
SODIUM SULFIDE RESIDUE: flammable, dangerous fire risk, strong irritant
to skin and tissue, incompatible with acids.

HAZARBOUS POLYMERIZATION: WILL NOT OCCUR

CONDITIONS TO AVOID: NA

₹4. HAZARDOUS INGREDIENTS (Mixtures Only)

MATERIAL OR COMPONENT/C.A.S. #: Not Applicable.

D. ENVIRONMENTAL

### DEGRADABILITY/AQUATIC TOXICITY:

Aquatic Toxicity: 2600 ppm/24, 48 & 96 hr/mosquito fish/TLm/fresh water. Biological Oxygen Demand (BOD): 0.12 lb/lb, instantaneous.—Reference (d). OCTANOL/WATER PARTITION COEFFICIENT: ND

EPA HAZARDOUS SUBSTANCES (CLEAN WATER ACT SEC. 311) ! NO

IF SO REPORTABLE QUANTITY: -- (40 CFR 116-117)

WASTE DISPOSAL METHODS (DISPOSER MUST COMPLY WITH FEDERAL, STATE AND LOCAL DISPOSAL OR DISCHARGE LAWS): Dissolve in water, using caution as solution can get hot. Neutralize with acid and flush to sewer with plenty of water if permitted by applicable disposal regulations. Good ventilation is required during neutralization due to release of SO2 gas. Oxidation to sodium sulfate solution may be required, as for example, by adding a slight excess of dilute hydrogen peroxide carefully and with stirring. Neutralized waste may have to be disposed of by an approved contractor.

RCRA STATUS OF UNUSED MATERIAL IF DISCARDED: Not a "hazardous waste". HAZARDOUS WASTE NUMBER: (IF APPLICABLE): NA (40 CFR 261)

## J. REFERENCES

PERMISSIBLE CONCENTRATION REFERENCES:

None.

REGULATORY STANDARDS:

FDA regulations apply to the use of food grade material (21 CFR). O.O.T. CLASSIFICATION: Not Regulated (49 CFR 173) GENERAL:

- (a) Monograph 139, "Sulfite Salts", Gosselin, R.E. et al., CLINICAL TOXICOLOGY OF COMMERCIAL PRODUCTS, 4th Ed., 1976.
- (b) NIOSH, REGISTRY OF TOXIC EFFECTS OF CHEMICAL SUBSTANCES, 1981-82, P881 154478, No. WE215 00 00.
- (c) NIOSH/OSHA "Pocket Guide to Chemical Hazards", DHHS (NIOSH) Pub. No. 78-210, 1978, Gov't. Printing Office, Washington 20402.
- (d) Coast Guard CHRIS system form SSF, "Sodium Sulfite", Oct. 1978.

### K. ADDITIONAL INFORMATION

Sodium sulfite has been demonstrated to be mutagenic in microbial systems;

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owever, it is not mutagenic in studies involving insects and is not onsidered to present a mutagenic threat to multicell organisms, (i.e. nimals, humans).

his product is not for drug or food use unless so labeled.

- or food grade product, the following applies:
- Effective July 9, 1986, the FDA has banned the use of "Sulfiting Agent" or "Sulfites" on fruits and vegetables intended to be served or sold raw to consumers.
- 2) Effective July 9, 1987, the FDA is requiring when a sulfite is present in a detectable amount in a finished food, regardless of whether it has been directly or indirectly added via one or more of the food ingredients, it must be declared on the label. The regulation defines a "detectable amount" of sulfite to be 10 ppm.
- 3) Sulfiting agents or sulfites are not to be used on foods or meats recognized as a source of Vitamin B1.

OOTNOTE: ND = NOT DETERMINED NA = NOT APPLICABLE

\* = PROPRIETARY - TRADE SECRET